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| (54) Title: RUTHENIUM AND OSMIUM METAL CARBENE COMPLEXES FOR OLEFIN METATHESIS POLYMERIZATION (57) Abstract Processes for the synthesis of several new carbene compounds of ruthenium and osmium are provided. These novel complexes function as stable, well-defined catalysts for the metathesis polymerisation of cyclic olefins. | | |

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TITLERUTHENIUM AND OSMIUM METAL CARBENE
COMPLEXES FOR OLEFIN METATHESIS POLYMERIZATION

5

BACKGROUND OF THE INVENTION

This invention relates to new ruthenium and osmium metal carbene complex compounds and their utility in an improved catalytic process for olefin metathesis polymerization.

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During the past two decades, research efforts have enabled an in depth understanding of the olefin metathesis reaction as catalyzed by early transition metal complexes. In contrast, the nature of the intermediates and the reaction mechanism for Group VIII transition metal catalysts has remained elusive. In particular, the oxidation states and ligation of the ruthenium and osmium metathesis intermediates are not known. Furthermore, the discrete ruthenium and osmium carbene complexes isolated to date do not exhibit metathesis activity.

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Many ruthenium and osmium metal carbenes have been reported in the literature (for example, see Burrell, A. K., Clark, G. R., Rickard, C. E. F., Roper, W. R., Wright, A. H., J. Chem. Soc., Dalton Trans., 1991, Issue 1, pp. 609-614).

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SUMMARY OF THE INVENTION

The present invention involves a reaction of a ruthenium or osmium compound with either a cyclopropene or a phosphorane to produce well-defined carbene compounds which can be called carbene complexes and which can catalyze the polymerization of cyclic olefin via ring-opening metathesis.

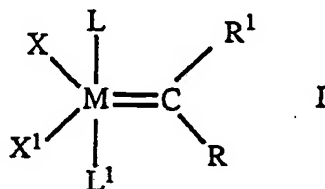
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The carbene compounds of the present invention are the only Ru and Os carbene complexes known to date in which the metal is formally in the +2 oxidation state,

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has an electron count of 16, and is pentacoordinate. The compounds claimed herein are active catalysts for ring-opening metathesis polymerization ("ROMP"). Most metathesis catalysts presently known are poisoned by functional groups and are, therefore, incapable of catalyzing metathesis polymerization reactions in protic or aqueous solvent systems.

Thus, the present invention pertains to compounds of the formula



wherein:

M is Os or Ru;

R and R¹ are independently selected from hydrogen; C₂-C₂₀ alkenyl, C₂-C₂₀ alkynyl, C₁-C₂₀ alkyl, aryl, C₁-C₂₀ carboxylate, C₁-C₂₀ alkoxy, C₂-C₂₀ alkenyloxy, C₂-C₂₀ alkynyloxy, aryloxy, C₂-C₂₀ alkoxycarbonyl, C₁-C₂₀ alkylthio, C₁-C₂₀ alkylsulfonyl or C₁-C₂₀ alkylsulfinyl; each optionally substituted with C₁-C₅ alkyl, halogen, C₁-C₅ alkoxy or with a phenyl group optionally substituted with halogen, C₁-C₅ alkyl or C₁-C₅ alkoxy;

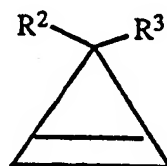
X and X¹ are independently selected from any anionic ligand; and

L and L¹ are independently selected from any neutral electron donor.

In one embodiment of these compounds, they can be in the form wherein 2, 3, or 4 of the moieties X, X¹, L, and L¹ can be taken together to form a chelating multidentate ligand. In one aspect of this embodiment,

X, L, and L¹ can be taken together to form a cyclopentadienyl, indenyl, or fluorenyl moiety.

The present invention also pertains to a method of preparing the aforementioned ruthenium and osmium compounds comprising reacting a compound of the formula $(XX^1ML_nL^1_m)_p$, in the presence of solvent, with a cyclopropene of the formula



wherein:

M, X, X^1 , L, and L^1 have the same meaning as indicated above;

n and m are independently 0-4, provided $n+m=2, 3$ or 4:

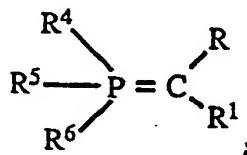
p is an integer equal to or greater than 1; and
R² and R³ are independently selected from hydrogen;

C₁-C₁₈ alkyl, C₂-C₁₈ alkenyl, C₂-C₁₈ alkynyl, C₂-C₁₈ alkoxy carbonyl, aryl, C₁-C₁₈ carboxylate, C₁-C₁₈ alkenyloxy, C₂-C₁₈ alkynyloxy, C₁-C₁₈ alkoxy, aryloxy, C₁-C₁₈ alkylthio, C₁-C₁₈ alkylsulfonyl or C₁-C₁₈ alkylsulfinyl; each optionally substituted with C₁-C₅ alkyl, halogen, C₁-C₅ alkoxy or with a phenyl group optionally substituted with halogen, C₁-C₅ alkyl or C₁-C₅ alkoxy.

In one embodiment of the process, X, L, and L¹ are taken together to form a moiety selected from the group consisting of cyclopentadienyl, indenyl or fluorenyl, each optionally substituted with hydrogen; C₂-C₂₀ alkenyl, C₂-C₂₀ alkynyl, C₁-C₂₀ alkyl, aryl, C₁-C₂₀ carboxylate, C₁-C₂₀ alkoxy, C₂-C₂₀ alkenyloxy, C₂-C₂₀

alkynyloxy, aryloxy, C₂-C₂₀ alkoxy carbonyl, C₁-C₂₀ alkylthio, C₁-C₂₀ alkylsulfonyl, C₁-C₂₀ alkylsulfinyl; each optionally substituted with C₁-C₅ alkyl, halogen, C₁-C₅ alkoxy or with a phenyl group optionally substituted with halogen, C₁-C₅ alkyl or C₁-C₅ alkoxy.

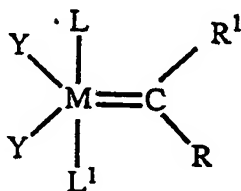
A still further method of preparing the compounds of this invention comprises reacting compound of the formula (XX¹ML_nL¹_m)_p in the presence of solvent with phosphorane of the formula



wherein:

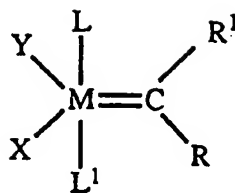
M, X, X¹, L, L¹, n, m, p, R, and R¹ have the same meaning as indicated above; and
 R⁴, R⁵ and R⁶ are independently selected from aryl, C₁-C₆ alkyl, C₁-C₆ alkoxy or phenoxy, each optionally substituted with halogen, C₁-C₃ alkyl, C₁-C₃ alkoxy, or with a phenyl group optionally substituted with halogen, C₁-C₅ alkyl or C₁-C₅ alkoxy.

Another embodiment of the invention comprises preparing compounds of Formulae II and III



II

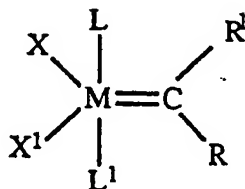
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III

from compound of Formula I

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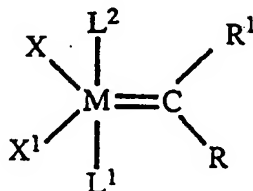
I

comprising reacting said compound of Formula I, in the
 10 presence of solvent, with compound of the formula M^1Y
 wherein:

M, R, R^1 , X, X^1 , L, and L^1 have the same meaning as
 indicated above, and wherein:

- 15 (1) M^1 is Li, Na or K, and Y is $\text{C}_1\text{-C}_{10}$ alkoxide or
 arylalkoxide each optionally substituted with
 $\text{C}_1\text{-C}_{10}$ alkyl or halogen, diaryloxide; or
 (2) M^1 is Na or Ag, and Y is ClO_4 , PF_6 , BF_4 , SbF_6 ,
 halogen, B(aryl)_4 , $\text{C}_1\text{-C}_{10}$ alkyl sulfonate or
 aryl sulfonate.

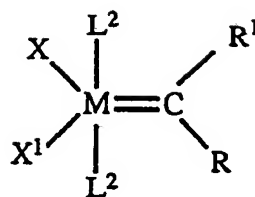
20 Another embodiment of the present invention is a
 method of preparing compounds of structures of Formulae
 IV and V



IV

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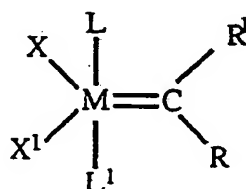
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V

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from compound of Formula I



10

I

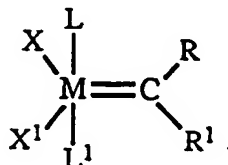
comprising reacting said compound I, in the presence of solvent, with L² wherein:

15 M, R, R¹, X, and X¹ have the same meaning as indicated above; and

L, L¹, and L² are independently selected from any neutral electron donor.

20 The compounds of Formulae II, III, IV, and V are species of, i.e., fall within, the scope of compounds of Formula I. In other words, certain compounds of Formula I are used to form by ligand exchange other compounds of Formula I. In this case, X and X¹ in Formula I are other than the Y in Formulae II and III that replaces X. Similarly, L and L¹ in Formula I are other than the L² 25 in Formulae IV and V. If any 2, 3, or 4 of X, X¹, L, and L¹ form a multidentate ligand of Formula I, only the remaining ligand moieties would be available for ligand replacement.

Still another embodiment of the present invention involves the use of compound I as a catalyst for polymerizing cyclic olefin. More specifically, this embodiment comprises metathesis polymerization of a
 5 polymerizable cyclic olefin in the presence of catalyst of the formula



10 in the presence of solvent, wherein:

M, R, R¹, X, X¹, L and L¹ have the same meaning as indicated above.

The reference above to X, X¹, L, and L¹ having the same meaning as indicated above refers to these moieties
 15 individually and taken together to form a multidentate ligand as described above.

DETAILED DESCRIPTION

The ruthenium and osmium metal complexes of the present invention are useful as catalysts in ring-
 20 opening metathesis polymerization, particularly in the living polymerization of strained cyclic olefins. Although all the criteria for a living polymer have not been completely established, the term living is used in the sense that the propagating moiety is stable and will
 25 continue to polymerize additional aliquots of monomer for a period after the original amount of monomer has been consumed. Aspects of this invention include the metal complex compounds, methods for their preparation, as well as their use as catalysts in the ROMP reaction.
 30 Uses for the resultant polymer are well documented in the book, *Olefin Metathesis*, by K. J. Ivin, Academic Press, Harcourt Brace Jovanovich Publishers (1983).

The intermediate compounds $(XX^1ML_nL^1_m)_p$ are either available commercially or can be prepared by standard known methods.

5 The phosphorane and cyclopropene reactants used in the present invention may be prepared in accordance with the following respective references. Schmidbaur, H. et al., *Phosphorus and Sulfur*, Vol. 18, pp. 167-170 (1983); Carter, F. L., Frampton, V. L., *Chemical Reviews*, Vol. 64, No. 5 (1964).

10 In the compounds of Formula I:

alkyl can include methyl, ethyl, *n*-propyl, *i*-propyl, or the several butyl, pentyl or hexyl isomers;

alkenyl can include 1-propenyl, 2-propenyl; 3-propenyl and the different butenyl, pentenyl and
15 hexenyl isomers, 1,3-hexadienyl and 2,4,6-heptatrienyl, and cycloalkenyl;

alkenyloxy can include $H_2C=CHCH_2O$, $(CH_3)_2C=CHCH_2O$, $(CH_3)CH=CHCH_2O$, $(CH_3)CH=C(CH_3)CH_2O$ and $CH_2=CHCH_2CH_2O$;

alkynyl can include ethynyl, 1-propynyl, 3-propynyl
20 and the several butynyl, pentynyl and hexynyl isomers, 2,7-octadiynyl and 2,5,8-decatriynyl;

alkynyloxy can include $HC\equiv CCH_2O$, $CH_3C\equiv CCH_2O$ and $CH_3C\equiv CCH_2OCH_2O$;

25 alkylthio can include, methylthio, ethylthio, and the several propylthio, butylthio, pentylthio and hexylthio isomers;

alkylsulfonyl can include CH_3SO_2 , $CH_3CH_2SO_2$, $CH_3CH_2CH_2SO_2$, $(CH_3)_2CHSO_2$ and the different butylsulfonyl, pentylsulfonyl and hexylsulfonyl isomers;

30 alkylsulfinyl can include CH_3SO , CH_3CH_2SO , $CH_3CH_2CH_2SO$, $(CH_3)_2CHSO$ and the different butylsulfinyl, pentylsulfinyl and hexylsulfinyl isomers;

carboxylate can include $CH_3CO_2CH_3CH_2CO_2$, $C_6H_5CO_2$, $(C_6H_5)CH_2CO_2$;

aryl can include phenyl, p-tolyl and p-fluorophenyl;

alkoxide can include methoxide, t-butoxide, and phenoxide;

5 diketonates can include acetylacetonate and 2,4-hexanedionate;

sulfonate can include trifluoromethanesulfonate, tosylate, and mesylate;

10 phosphine can include trimethylphosphine, triphenylphosphine, and methyldiphenylphosphine;

phosphite can include trimethylphosphite, triphenylphosphite, and methyldiphenylphosphite;

phosphinite can include triphenylphosphinite, and methyldiphenylphosphinite;

15 arsine can include triphenylarsine and trimethylarsine;

stibine can include triphenylstibine and trimethylstibine;

20 amine can include trimethylamine, triethylamine and dimethylamine;

ether can include $(\text{CH}_3)_3\text{CCH}_2\text{OCH}_2\text{CH}_3$, THF, $(\text{CH}_3)_3\text{COC}(\text{CH}_3)_3$, $\text{CH}_3\text{OCH}_2\text{CH}_2\text{OCH}_3$, and $\text{CH}_3\text{OC}_6\text{H}_5$;

thioether can include CH_3SCH_3 , $\text{C}_6\text{H}_5\text{SCH}_3$, $\text{CH}_3\text{OCH}_2\text{CH}_2\text{SCH}_3$, and tetrahydrothiophene;

25 amide can include $\text{HC}(=\text{O})\text{N}(\text{CH}_3)_2$ and $(\text{CH}_3)\text{C}(=\text{O})\text{N}(\text{CH}_3)_2$;

sulfoxide can include $\text{CH}_3\text{S}(=\text{O})\text{CH}_3$, $(\text{C}_6\text{H}_5)_2\text{SO}$;

30 alkoxy can include methoxy, ethoxy, n-propyloxy, isopropyloxy and the different butoxy, pentoxy and hexyloxy isomers, cycloalkoxy can include cyclopentyloxy and cyclohexyloxy;

cycloalkyl can include cyclopropyl, cyclobutyl, cyclopentyl, and cyclohexyl; and

35 cycloalkenyl can include cyclopentenyl and cyclohexenyl.

The term "halogen" or "halide", either alone or in compound words such as "haloalkyl", denotes fluorine, chlorine, bromine or iodine.

Alkoxyalkyl can include CH_3OCH_2 , $\text{CH}_3\text{OCH}_2\text{CH}_2$,
5 $\text{CH}_3\text{CH}_2\text{OCH}_2$, $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OCH}_2$ and $\text{CH}_3\text{CH}_2\text{OCH}_2\text{CH}_2$; and
alkoxycarbonyl can include $\text{CH}_3\text{OC}(=\text{O})$, $\text{CH}_3\text{CH}_2\text{OC}(=\text{O})$,
 $\text{CH}_3\text{CH}_2\text{CH}_2\text{OC}(=\text{O})$, $(\text{CH}_3)_2\text{CHOC}(=\text{O})$ and the different
butoxy-, pentoxy- or hexyloxycarbonyl isomers.

A neutral electron donor is any ligand which, when
10 removed from a metal center in its closed shell electron
configuration, has a neutral charge, i.e., is a Lewis
base.

An anionic ligand is any ligand which when removed
from a metal center in its closed shell electron
15 configuration has a negative charge. The critical
feature of the carbene compounds of this invention is
the presence of the ruthenium or osmium in the +2
oxidation state, an electron count of 16 and
pentacoordination. A wide variety of ligand moieties X,
20 X^1 , L, and L^1 can be present and the carbene compound
will still exhibit its catalytic activity.

A preferred embodiment of the compounds of the
present invention is:

A compound of the invention of Formula I wherein:

25 R and R^1 are independently selected from
hydrogen, vinyl, C_1 - C_{10} alkyl, aryl, C_1 - C_{10}
carboxylate, C_2 - C_{10} alkoxy carbonyl, C_1 - C_{10}
alkoxy, aryloxy, each optionally substituted
with C_1 - C_5 alkyl, halogen, C_1 - C_5 alkoxy or
30 with a phenyl group optionally substituted
with halogen, C_1 - C_5 alkyl or C_1 - C_5 alkoxy;
X and X^1 are independently selected from
halogen, hydrogen; or C_1 - C_{20} alkyl, aryl,
 C_1 - C_{20} alkoxide, aryloxy, C_2 - C_{20}
35 alkoxy carbonyl, arylcarboxylate, C_1 - C_{20}

carboxylate, aryl or C₁-C₂₀ alkylsulfonate,
C₁-C₂₀ alkylthio, C₁-C₂₀ alkylsulfonyl,
C₁-C₂₀ alkylsulfinyl, each optionally
substituted with C₁-C₅ alkyl, halogen, C₁-C₅
alkoxy or with a phenyl group optionally
substituted with halogen, C₁-C₅ alkyl or
C₁-C₅ alkoxy; and

L and L¹ are independently selected from
phosphine, sulfonated phosphine, phosphite,
phosphinite, phosphonite, arsine, stibine,
ether, amine, amide, sulfoxide, carbonyl,
nitrosyl, pyridine or thioether.

A more preferred embodiment of Formula I comprises:

A compound of the invention wherein:

R and R¹ are independently selected from
hydrogen; vinyl, C₁-C₅ alkyl, phenyl,
C₂-C₅ alkoxy carbonyl, C₁-C₅ carboxylate,
C₁-C₅ alkoxy, phenoxy; each optionally
substituted with C₁-C₅ alkyl, halogen,
C₁-C₅ alkoxy or a phenyl group
optionally substituted with halogen,
C₁-C₅ alkyl or C₁-C₅ alkoxy;

X and X¹ are independently selected from Cl,
Br, H, or benzoate, C₁-C₅ carboxylate,
C₁-C₅ alkyl, phenoxy, C₁-C₅ alkoxy, C₁-C₅
alkylthio, aryl, and C₁-C₅ alkyl
sulfonate; each optionally substituted
with C₁-C₅ alkyl or a phenyl group
optionally substituted with halogen,
C₁-C₅ alkyl or C₁-C₅ alkoxy;

L and L¹ are independently selected from
aryl or C₁-C₁₀ alkylphosphine, aryl- or
C₁-C₁₀ alkylsulfonated phosphine, aryl-
or C₁-C₁₀ alkylphosphinite, aryl- or
C₁-C₁₀ alkylphosphonite, aryl- or C₁-C₁₀

alkylphosphite, aryl- or C₁-C₁₀
 alkylarsine, aryl- or C₁-C₁₀ alkylamine,
 pyridine, aryl- or C₁-C₁₀ alkyl
 sulfoxide, aryl- or C₁-C₁₀ alkylether,
 5 or aryl- or C₁-C₁₀ alkylamide, each
 optionally substituted with a phenyl
 group optionally substituted with
 halogen, C₁-C₅ alkyl or C₁-C₅ alkoxy.

A further preferred embodiment of Formula I
 10 comprises:

A compound of the present invention wherein:

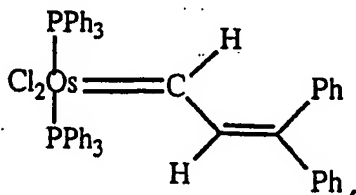
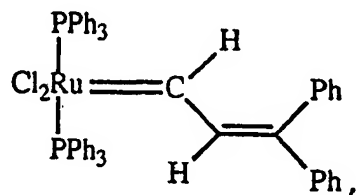
R and R¹ are independently vinyl, H, Me, Ph;
 X and X¹ are independently Cl, CF₃CO₂, CH₃CO₂
 CFH₂CO₂, (CH₃)₃CO, (CF₃)₂(CH₃)CO,
 15 (CF₃)(CH₃)₂CO, PhO, MeO, EtO, tosylate,
 mesylate, or trifluoromethanesulfonate;
 and

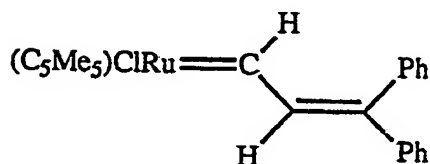
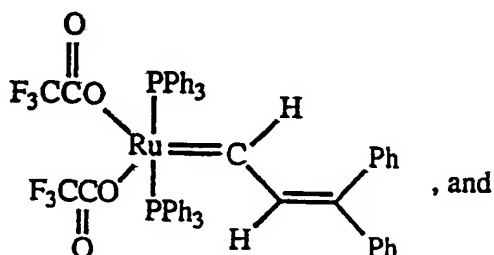
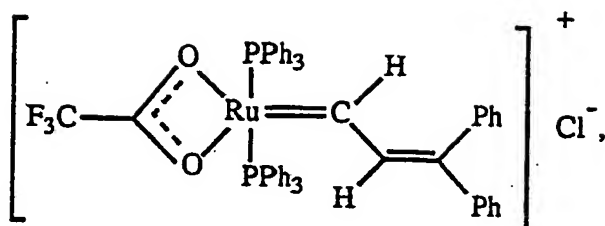
L and L¹ are independently PMe₃, PPh₃,
 P(p-Tol)₃, P(o-Tol)₃, PMePh₂, PPhMe₂,
 20 P(CF₃)₃, P(p-FC₆H₄)₃, pyridine,
 P(p-CF₃C₆H₄)₃, (p-F)pyridine,
 (p-CF₃)pyridine, P(C₆H₄-SO₃Na)₃ or
 P(CH₂C₆H₄-SO₃Na)₃.

For any of the foregoing described preferred groups
 25 of compounds, any 2, 3, or 4 of X, X¹, L, L¹ can be
 taken together to form a chelating multidentate ligand.
 Examples of bidentate ligands include, but are not
 limited to, bisphosphines, dialkoxides,
 alkylidicketonates, and arylidicketonates. Specific
 30 examples include Ph₂PCH₂CH₂PPh₂, Ph₂AsCH₂CH₂AsPh₂,
 Ph₂PCH₂CH₂C(CF₃)O⁻, binaphtholate dianions, pinacolate
 dianions, Me₂P(CH₂)₂PMe₂ and ⁻OC(CH₃)₂(CH₃)₂CO⁻.
 Preferred bidentate ligands are Ph₂PCH₂CH₂PPh₂ and
 Me₂PCH₂CH₂PMe₂. Tridentate ligands include, but are not
 35 limited to, (CH₃)₂NCH₂CH₂P(Ph)CH₂CH₂N(CH₃)₂. Other

preferred tridentate ligands are those in which X, L, and L¹ are taken together to be cyclopentadienyl, indenyl or fluorenyl, each optionally substituted with C₂-C₂₀ alkenyl, C₂-C₂₀ alkynyl, C₁-C₂₀ alkyl, aryl, C₁-C₂₀ carboxylate, C₁-C₂₀ alkoxy, C₂-C₂₀ alkenyloxy, C₂-C₂₀ alkynyloxy, aryloxy, C₂-C₂₀ alkoxycarbonyl, C₁-C₂₀ alkylthio, C₁-C₂₀ alkylsulfonyl, C₁-C₂₀ alkylsulfinyl, each optionally substituted with C₁-C₅ alkyl, halogen, C₁-C₅ alkoxy or with a phenyl group optionally substituted with halogen, C₁-C₅ alkyl or C₁-C₅ alkoxy. More preferably in compounds of this type, X, L, and L¹ are taken together to be cyclopentadienyl or indenyl, each optionally substituted with hydrogen; vinyl, C₁-C₁₀ alkyl, aryl, C₁-C₁₀ carboxylate, C₂-C₁₀ alkoxycarbonyl, C₁-C₁₀ alkoxy, aryloxy, each optionally substituted with C₁-C₅ alkyl, halogen, C₁-C₅ alkoxy or with a phenyl group optionally substituted with halogen, C₁-C₅ alkyl or C₁-C₅ alkoxy. Most preferably, X, L, and L¹ are taken together to be cyclopentadienyl, optionally substituted with vinyl, hydrogen, Me or Ph. Tetradentate ligands include, but are not limited to O₂C(CH₂)₂P(Ph)(CH₂)₂P(Ph)(CH₂)₂CO₂, phthalocyanines, and porphyrins.

The most preferred carbene compounds of the present invention include:

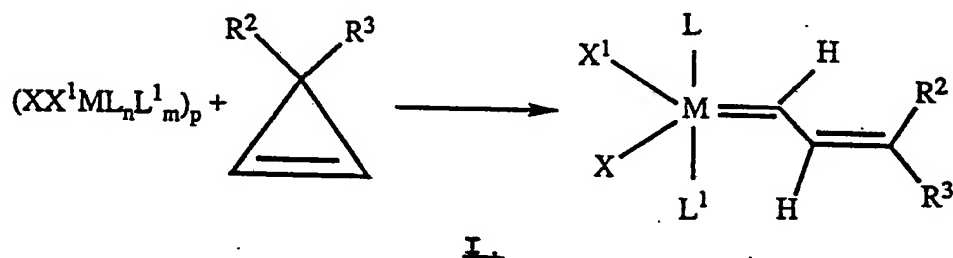




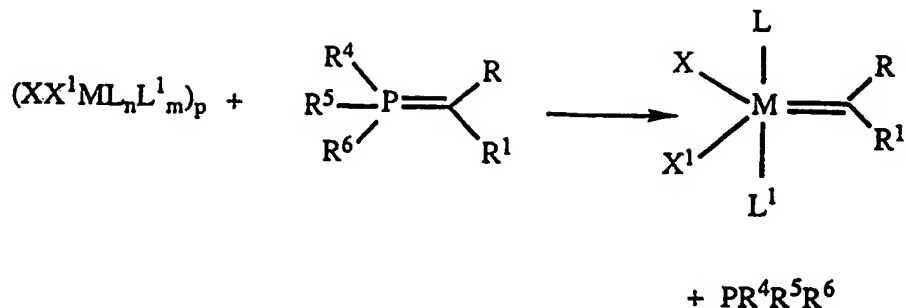
The compounds of the present invention can be prepared in several different ways, each of which is described below.

The most general method for preparing the compounds of this invention comprises reacting $(\text{XX}^1\text{ML}_n\text{L}^1_m)_p$ with a cyclopropene or phosphorane in the presence of a solvent to produce a carbene complex, as shown in the equations.

REACTION EQUATIONS



15

II.

wherein:

M, X, X¹, L, L¹, n, m, p, R², R³, R⁴, R⁵, and R⁶ are as defined above. Preferably, R², R³, R⁴, R⁵, and R⁶ are independently selected from the group consisting of C₁-C₆ alkyl or phenyl.

Examples of solvents for this reaction include organic, protic, or aqueous solvents which are inert under the reaction conditions, such as: aromatic hydrocarbons, chlorinated hydrocarbons, ethers, aliphatic hydrocarbons, alcohols, water, or mixtures thereof. Preferred solvents include benzene, toluene, *p*-xylene, methylene chloride, dichloroethane, dichlorobenzene, tetrahydrofuran, diethylether, pentane, methanol, ethanol, water, or mixtures thereof. More preferably, the solvent is benzene, toluene, *p*-xylene, methylene chloride, dichloroethane, dichlorobenzene, tetrahydrofuran, diethylether, pentane, methanol, ethanol, or mixtures thereof.

A suitable temperature range is from about -20°C to about 125°C, preferably 35°C to 90°C, and more preferably 50°C to 65°C. Pressure is not critical but may depend on the boiling point of the solvent used, i.e., use sufficient pressure to maintain a solvent liquid phase. Reaction times are not critical, and can be from several minutes to 48 hours. The reactions are

generally carried out in an inert atmosphere, most preferably nitrogen or argon.

The reaction is usually carried out by dissolving the compound $(XX^1ML_nL^1_m)_p$, in a suitable solvent, adding the cyclopropene (preferably in a solvent) to a stirred solution of the compound, and optionally heating the mixture until the reaction is complete. The progress of the reaction can be monitored by any of several standard analytical techniques, such as infrared or nuclear magnetic resonance. Isolation of the product can be accomplished by standard procedures, such as evaporating the solvent, washing the solids (e.g., with alcohol or benzene), and then recrystallizing the desired carbene complex. Whether the moieties X, X^1 , L, or L^1 are (unidentate) ligands or some taken together to form multidentate ligands will depend on the starting compound which simply carries these ligands over into the desired carbene complex.

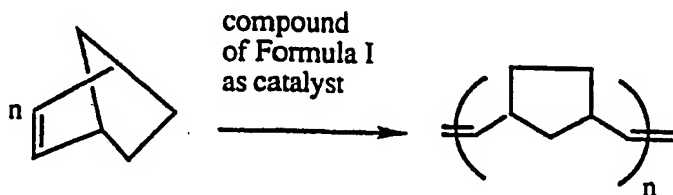
In one variation of this general procedure, the reaction is conducted in the presence of $HgCl_2$, preferably 0.01 to 0.2 molar equivalents, more preferably 0.05 to 0.1 equivalents, based on $XX^1ML_nL^1_m$. In this variation, the reaction temperature is preferably 15°C to 65°C.

In a second variation of the general procedure, the reaction is conducted in the presence of ultraviolet radiation. In this variation, the reaction temperature is preferably -20°C to 30°C.

It is also possible to prepare carbene complexes of this invention by ligand exchange. For example, L and/or L^1 can be replaced by a neutral electron donor, L^2 , in compounds of Formula I by reacting L^2 with compounds of Formula I wherein L, L^1 , and L^2 are independently selected from phosphine, sulfonated phosphine, phosphite, phosphinite, phosphonite, arsine,

stibine, ether, amine, amide, sulfoxide, carbonyl, nitrosyl, pyridine or thioether. Similarly, X and/or X¹ can be replaced by an anionic ligand, Y, in compounds of Formula I by reacting M¹Y with compounds of Formula I, wherein X and X¹ are independently selected from halogen, hydrogen, or C₁-C₂₀ alkyl, aryl, C₁-C₂₀ alkoxide, aryloxy, C₂-C₂₀ alkoxy carbonyl, arylcarboxylate, C₁-C₂₀ carboxylate, aryl or C₁-C₂₀ alkylsulfonate, C₁-C₂₀ alkylthio, C₁-C₂₀ alkylsulfonyl, C₁-C₂₀ alkylsulfinyl, each optionally substituted with C₁-C₅ alkyl, halogen, C₁-C₅ alkoxy or with a phenyl group optionally substituted with halogen, C₁-C₅ alkyl or C₁-C₅ alkoxy. These ligand exchange reactions are typically carried out in a solvent which is inert under the reaction conditions. Examples of solvents include those described above for the preparation of the carbene complex.

The compounds of this invention are useful as catalysts in the preparation of a wide variety of polymers which can be formed by ring-opening metathesis polymerization of cyclic olefins. Therefore, one embodiment of this invention is an improved polymerization process comprising metathesis polymerization of a cyclic olefin, wherein the improvement comprises conducting the polymerization in the presence of a catalytic amount of a compound of Formula I. The polymerization reaction is exemplified for norbornene in the following equation:



wherein n is the repeat unit of the polymeric chain.

Examples of cyclic olefins for this polymerization process include norbornene, norbornadiene, cyclopentene, dicyclopentadiene, cycloheptene, cyclo-octene, 7-oxanorbornene, 7-oxanorbornadiene, and cyclododecene.

5 The polymerization reaction is generally carried out in an inert atmosphere by dissolving a catalytic amount of a compound of Formula I in a solvent and adding the cyclic olefin, optionally dissolved in a solvent, to the catalyst solution. Preferably, the
10 reaction is agitated (e.g., stirred). The progress of the reaction can be monitored by standard techniques, e.g., nuclear magnetic resonance spectroscopy.

 Examples of solvents for the polymerization reaction include organic, protic, or aqueous solvents
15 which are inert under the polymerization conditions, such as: aromatic hydrocarbons, chlorinated hydrocarbons, ethers, aliphatic hydrocarbons, alcohols, water, or mixtures thereof. Preferred solvents include benzene, toluene, p-xylene, methylene chloride,
20 dichloroethane, dichlorobenzene, tetrahydrofuran, diethylether, pentane, methanol, ethanol, water, or mixtures thereof. More preferably, the solvent is benzene, toluene, p-xylene, methylene chloride, dichloroethane, dichlorobenzene, tetrahydrofuran,
25 diethylether, pentane, methanol, ethanol, or mixtures thereof. Most preferably, the solvent is toluene or a mixture of benzene and methylene chloride. The solubility of the polymer formed in the polymerization reaction will depend on the choice of solvent and the
30 molecular weight of the polymer obtained.

 Reaction temperatures can range from 0°C to 100°C, and are preferably 25°C to 45°C. The ratio of catalyst to olefin is not critical, and can range from 1:5 to 1:10,000, preferably 1:10 to 1:1,000.

Because the compounds of Formula I are stable in the presence of protic solvents, the polymerization reaction may also be conducted in the presence of a protic solvent. This is very unusual among metathesis catalysts and provides a distinct advantage for the process of this invention over the processes of the prior art. Other advantages of the polymerization process of this invention derive from the fact that the compounds of Formula I are well-defined, stable Ru or Os carbene complexes providing high catalytic activity. Using such compounds as catalysts allows control of the rate of initiation, extent of initiation, and the amount of catalyst. Also, the well-defined ligand environment of these complexes provides flexibility in modifying and fine-tuning their activity level, solubility, and stability. In addition, these modifications enable ease of recovery of catalyst.

General Description of the Preparation of Compounds of this Invention from Cyclopropenes:

A 50 ml Schlenk flask equipped with a magnetic stirbar is charged with $(\text{MXX}^{\text{L}}\text{L}_n\text{L}^{\text{L}}_m)_p$ (0.1 mmol) inside a nitrogen-filled drybox. Methylene chloride (2 ml) is added to dissolve the complex followed by 25 ml of benzene to dilute the solution. One equivalent of a cyclopropene is then added to this solution. The reaction flask is then capped with a stopper, removed from the box, attached to a reflux condenser under argon and heated at 55°C. The reaction is then monitored by NMR spectroscopy until all the reactants have been converted to product. At the end of the reaction, the solution is allowed to cool to room temperature under argon and then filtered into another Schlenk flask via a cannula filter. All solvent is then removed *in vacuo* to give a solid. This solid is then washed with a solvent

in which the by-product will be soluble but the desired product will not. After the washing supernatant is removed, the resulting solid powder is dried *in vacuo* overnight. Further purification via crystallization can be performed if necessary.

The abbreviations Me, Ph, and THF used herein refer to methyl, phenyl, and tetrahydrofuran, respectively.

Representative compounds of the present invention which are prepared in accordance with the procedure described above are exemplified in Table I.

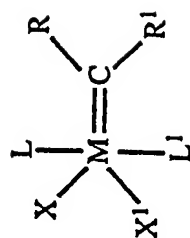
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25

30

TABLE I



| Compound Name | M | X | X ¹ | L | L ¹ | R | R ¹ |
|---|----|----|----------------|---------------------|---------------------|---|---------------------|
| Dichloro-3,3-diphenylvinyl-carbene-bis(triphenylphosphine) ruthenium(II) | Ru | Cl | Cl | PPh ₃ | PPh ₃ | H | CH=CPH ₂ |
| Dibromo-3,3-diphenylvinyl-carbene-bis(triphenylphosphine) ruthenium(II) | Ru | Br | Br | PPh ₃ | PPh ₃ | H | CH=CPH ₂ |
| Dichloro-3,3-diphenylvinyl-carbene-bis(methyldiphenylphosphine) ruthenium(II) | Ru | Cl | Cl | PPh ₂ Me | PPh ₂ Me | H | CH=CPH ₂ |
| Dibromo-3,3-diphenylvinyl-carbene-bis(methyldiphenylphosphine) ruthenium(II) | Ru | Br | Br | PPh ₂ Me | PPh ₂ Me | H | CH=CPH ₂ |

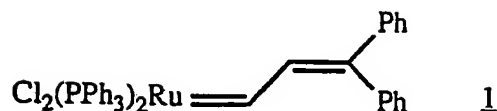
| Compound Name | M | X | X ¹ | L | L ¹ | R | R ¹ |
|--|----|----|----------------|------------------|------------------|---|----------------|
| Dichloro-3-methyl-3-phenylvinylcarbene-bis(triphenylphosphine)-ruthenium(II) | Ru | Cl | Cl | PPh ₃ | PPh ₃ | H | |
| Dibromo-3-methyl-3-phenylvinylcarbene-bis(triphenylphosphine)-ruthenium(II) | Ru | Br | Br | PPh ₃ | PPh ₃ | H | |
| Dichloro-3,3-dimethyl-3-phenylvinylcarbene-bis(triphenylphosphine)-ruthenium(II) | Ru | Cl | Cl | PPh ₃ | PPh ₃ | H | |
| Bis(acetato)-3,3-diphenyl-3-phenylvinylcarbene-bis(triphenylphosphine)-ruthenium(II) | Ru | | | PPh ₃ | PPh ₃ | H | |
| Acetato-3,3-diphenyl-3-phenylvinylcarbene-bis(triphenylphosphine)-ruthenium(II)-chloride | Ru | | Cl | PPh ₃ | PPh ₃ | H | |
| 3,3-Diphenylvinylcarbene-bis(trifluoroacetato)-bis(triphenylphosphine)-ruthenium(II) | Ru | | | PPh ₃ | PPh ₃ | H | |

| Compound Name | M | X | X ¹ | L | L ¹ | R | R ¹ |
|---|----|--------------------|----------------|------------------|------------------|---|----------------|
| 3,3-Diphenylvinylcarbene-n ² -pinacol-bis(triphenylphosphine) ruthenium(II) | Ru | | | PPh ₃ | PPh ₃ | H | |
| 3,3-Diphenylvinylcarbene-bis(t-butoxy)bis-(triphenylphosphine) ruthenium(II) | Ru | Me ₃ CO | | PPh ₃ | PPh ₃ | H | |
| 3,3-Diphenylvinylcarbene-bis(2-trifluoromethyl-2-propoxy)-bis(triphenylphosphine) ruthenium(II) | Ru | | | PPh ₃ | PPh ₃ | H | |

These are representative examples of the ruthenium complexes. Analogous complexes could be made with osmium.

EXAMPLE I

Synthesis of



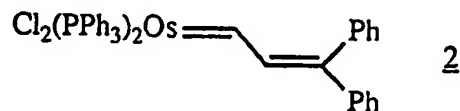
- 5 In a typical reaction, a 200 ml Schlenk flask equipped with a magnetic stirbar was charged with $\text{RuCl}_2(\text{PPh}_3)_4$ (6.00 g, 4.91 mmol) inside a nitrogen-filled drybox. Methylene chloride (40 mL) was added to
- 10 dissolve the complex followed by 100 mL of benzene to dilute the solution. 3,3-Diphenylcyclopropene (954 mg, 1.01 equiv) was then added to the solution via pipette. The reaction flask was capped with a stopper, removed from the box, attached to a reflux condenser under argon
- 15 and heated at 53°C for 11 h. After allowing the solution to cool to room temperature, all the solvent was removed in vacuo to give a dark yellow-brown solid. Benzene (10 mL) was added to the solid and subsequent swirling of the mixture broke the solid into a fine
- 20 powder. Pentane (80 mL) was then slowly added to the mixture via cannula while stirring vigorously. The mixture was stirred at room temperature for 1 h and allowed to settle before the supernatant was removed via cannula filtration. This washing procedure was repeated
- 25 two more times to ensure the complete removal of all phosphine by-products. The resulting solid was then dried under vacuum overnight to afford 4.28 g (98%) of Compound 1 as a yellow powder with a slight green tint. ^1H NMR (C_6D_6): δ 17.94 (pseudo-quartet = two
- 30 overlapping triplets, 1H, $\text{Ru}=\text{CH}$, $J_{\text{HH}}=10.2$ Hz, $J_{\text{PH}}=9.7$ Hz), 8.33 (d, 1H, $\text{CH}=\text{CPh}_2$, $J_{\text{HH}} 10.2$ Hz). ^{31}P NMR (C_6D_6): δ 28.2 (s). ^{13}C NMR (CD_2Cl_2): δ 288.9 (t, M = C, $J_{\text{CP}}=10.4$ Hz), 149.9 (t, $\text{CH}=\text{CPh}_2$, $J_{\text{CP}}=11.58$ Hz).

The carbene complex which is the compound formed in the above example is stable in the presence of water or alcohol.

EXAMPLE II

5

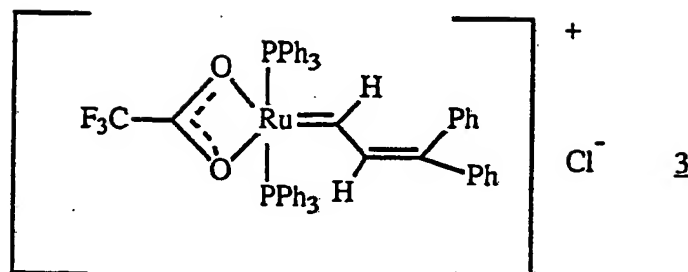
Synthesis procedure for



A 50 ml Schlenk flask equipped with a magnetic stirbar was charged with $\text{OsCl}_2(\text{PPh}_3)_3$ (100 mg, 0.095 mmol) inside a nitrogen-filled drybox. Methylene chloride (2 ml) was added to dissolve the complex followed by 25 ml of benzene to dilute the solution. 3,3-diphenylcyclopropene (18.53 mg, 1.01 eq) was then added to the solution via pipet. The reaction flask was capped with a stopper, removed from the box, attached to a reflux condenser under argon and heated at 55°C for 14 h. After allowing the solution to cool to room temperature, all the solvent was removed *in vacuo* to give a dark yellow-brown solid. Benzene (2 ml) was added to the solid and subsequent swirling of the mixture broke the solid into a fine powder. Pentane (30 ml) was then slowly added to the mixture via cannula while stirring vigorously. The mixture was stirred at RT for 1 h and allowed to settle before the supernatant was removed via cannula filtration. This washing procedure was repeated two more times to ensure the complete removal of all phosphine by-products. The resulting solid was then dried under vacuum overnight to afford 74.7 mg of Compound 2 as a yellow powder (80%).
 ^1H NMR (C_6D_6): δ 19.89 (pseudo-quartet = two overlapping triplets, 1H, Os = CH, $J_{\text{HH}} = 10.2$ Hz), 8.23 (d, 1H, CH=CPh₂, $J_{\text{HH}} = 10.2$ Hz). ^{31}P NMR (C_6D_6): δ 4.98 (s).

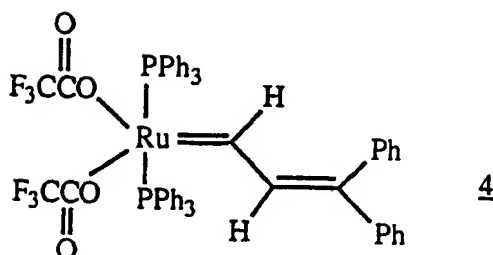
EXAMPLE III

Synthesis of



- 5
- A 50 ml Schlenk flask equipped with a magnetic stirbar was charged with $\text{RuCl}_2(\text{PPh}_3)_2(=\text{CH}-\text{CH}=\text{CPh}_2)$ (100 mg, 0.18 mmol) inside a nitrogen-filled drybox. Methylene chloride (10 ml) was added to dissolve the complex. AgCF_3CO_2 (24.9 mg., 1 eq) was weighed into a
- 10 10 ml round-bottom flask, dissolved with 3 ml of THF. Both flasks were then capped with rubber septa and removed from the box. The Schlenk flask was then put under an argon atmosphere and the AgCF_3CO_2 solution was
- 15 added dropwise to this solution via a gas-tight syringe over a period of 5 min while stirring. At the end of the addition, there was a lot of precipitate in the reaction mixture and the solution turned into a fluorescent green color. The supernatant was
- 20 transferred into another 50 ml Schlenk flask under argon atmosphere via the use of a cannula filter. Subsequent solvent removal under *in vacuo* and washing with pentane (10 ml) afforded a green solid powder, Compound 3. Yield = 92.4 mg (85%).
- 25 ^1H NMR (2:2:1 CD_2Cl_2 : C_6D_6 :THF- d_8) : δ 18.77 (dt, 1H, $\text{Ru}=\text{CH}$, $J_{\text{HH}}=11.2$ Hz, $J_{\text{PH}}=8.6$ Hz), 8.40 (d, 1H), $\text{CH}=\text{CPh}_2$, $J_{\text{HH}}=11.2$ Hz). ^{31}P NMR (2:2:1 CD_2Cl_2 : C_6D_6 :THF- d_8) δ 29.4. ^{19}F NMR (2:2:1 CD_2Cl_2 : C_6D_6 :THF- d_8) δ 75.8.

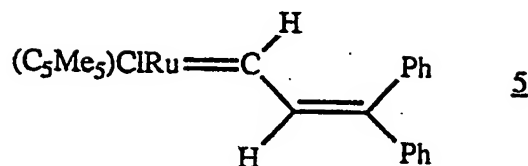
EXAMPLE IV
Synthesis of



- 5
- A 50 ml Schlenk flask equipped with a magnetic stirbar was charged with $\text{RuCl}_2(\text{PPh}_3)_2(=\text{CH}-\text{CH}=\text{CPh}_2)$ (100 mg, 0.11 mmol) inside a nitrogen-filled drybox. Methylene chloride (10 ml) was added to dissolve the complex. AgCF_3CO_2 (49.8 mg, 2 eq) was weighed into a 10 ml round-bottom flask, dissolved with 4 ml of THF. Both
- 10 flasks were then capped with rubber septa and removed from the box. The Schlenk flask was then put under an argon atmosphere and the AgCF_3CO_2 solution was added
- 15 dropwise via a gas tight syringe over a period of 5 min to the solution of ruthenium compound while stirring. At the end of the addition, there was a lot of precipitate in the reaction mixture and the solution
- 20 turned into a fluorescent lime green color. The supernatant was transferred into another 50 ml Schlenk flask under argon atmosphere with the use of a cannula filter. Subsequent solvent removal *in vacuo* and washing with pentane (10 ml) afforded a green powder, Compound
4. Yield = 102 mg (87%).
- 25 ^1H NMR (2:2:1 CD_2Cl_2 : C_6D_6 :THF- d_8) δ 19.23 (dt, slightly overlapping) $\text{Ru}=\text{CH}$, $J_{\text{HH}}=11.5$ Hz, $J_{\text{PH}}=5.4$ Hz), 8.07 (d, 1H, $\text{CH}=\text{CPh}_2$, $J_{\text{HH}}=11.5$ Hz). ^{31}P NMR (2:2:1 CD_2Cl_2 : C_6D_6 :THF- d_8) δ 28.6. ^{19}F NMR (2:2:1 CD_2Cl_2 : C_6D_6 :THF- d_8) δ -75.7.

EXAMPLE V

Synthesis of



5 The reaction between $[\text{Ru}(\text{C}_5\text{Me}_5)\text{Cl}]_4$ and
 3,3-diphenylcyclopropene was done under a nitrogen
 atmosphere. $[\text{Ru}(\text{C}_5\text{Me}_5)\text{Cl}]_4$ (100 mg, 0.092 mmol) was
 dissolved in 10 mL of tetrahydrofuran. To this solution
 10 was added 3,3-diphenylcyclopropene (350 mg, 1.82 mmol).
 The resulting solution was stirred at room temperature
 for 1 h. Petroleum ether (10 mL) was then added to the
 reaction mixture. It was stirred for an additional
 30 min, and then all volatile components were removed
 15 from the reaction mixture under vacuum. The crude
 product was extracted with diethyl ether; volatiles were
 removed from the filtrate under vacuum to afford a dark
 colored, oily solid. This was further extracted with
 petroleum ether; volatiles were removed from the
 20 filtrate under vacuum to afford a very dark red-brown
 oil. This was recrystallized from petroleum ether at
 -40°C to afford dark crystals. NMR spectra of which are
 consistent with the formulation $[\text{Ru}(\text{C}_5\text{Me}_5)(\text{CHC}=\text{CPh}_2)\text{Cl}]_n$
 (value of n as yet undetermined: e.g., the product
 25 could be a dimer).

EXAMPLE VI

Polymerization of Norbornene
 Using Compound of Example 1

30 $(\text{PPh}_3)_2\text{Cl}_2\text{Ru}=\text{CH}-\text{CH}=\text{CPh}_2$ catalyzed polymerized
 norbornene in a 1:8 mixture of $\text{CH}_2\text{Cl}_2/\text{C}_6\text{H}_6$ at room
 temperature to yield polynorbornene. A new signal,

attributed to H_α of the propagating carbene, was observed by 1H NMR spectroscopy at 17.79 ppm. Its identity and stability was confirmed by preparing a block polymer with 2,3-dideuteronorbornene and perprotonorbornene. When 2,3-dideuteronorbornene was added to the propagating species, the new carbene signal vanished and then reappeared when perprotonorbornene was added for the third block.

EXAMPLE VII

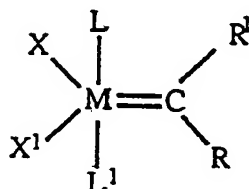
Polymerization of Norbornene Using Compound of Example 5

$Ru(C_5Me_5)(CHC=CPh_2)Cl$ (14 mg, 0.030 mmol) was dissolved in 1 mL of perdeuterated toluene under a nitrogen atmosphere. To this was added norbornene (109 mg, 1.16 mmol). The reaction mixture became viscous within minutes as the norbornene polymerized. After 20 hrs at room temperature a 1H NMR spectrum of the reaction mixture was recorded, which showed polynorbornene and unreacted norbornene monomer in a ratio of 82:12.

CLAIMS

What is claimed is

- 5 1. A compound of the formula



wherein:

- 10 M is Os or Ru;
 R and R¹ are independently selected from hydrogen;
 C₂-C₂₀ alkenyl, C₂-C₂₀ alkynyl, C₁-C₂₀ alkyl,
 aryl, C₁-C₂₀ carboxylate, C₁-C₂₀ alkoxy, C₂-C₂₀
 alkenyloxy, C₂-C₂₀ alkynyloxy, aryloxy, C₂-C₂₀
 15 alkoxycarbonyl, C₁-C₂₀ alkylthio, C₁-C₂₀
 alkylsulfonyl or C₁-C₂₀ alkylsulfinyl; each
 optionally substituted with C₁-C₅ alkyl,
 halogen, C₁-C₅ alkoxy or with a phenyl group
 optionally substituted with halogen, C₁-C₅ alkyl
 20 or C₁-C₅ alkoxy;
 X and X¹ are independently selected from any anionic
 ligand; and
 L and L¹ are independently selected from any neutral
 electron donor.

25

2. A compound according to Claim 1 wherein any 2,
 3 or 4 of X, X¹, L, and L¹ are bonded together to form a
 chelating multidentate ligand.

30

3. A compound according to Claim 2 wherein 2 of
 X, X¹, L, and L¹ are bonded together to form a bidentate
 ligand.

4. A compound according to Claim 2 wherein X, L and L¹ are taken together to be cyclopentadienyl, indenyl or fluorenyl, each optionally substituted with hydrogen; C₂-C₂₀ alkenyl, C₂-C₂₀ alkynyl, C₁-C₂₀ alkyl, aryl, C₁-C₂₀ carboxylate, C₁-C₂₀ alkoxy, C₂-C₂₀ alkenyloxy, C₂-C₂₀ alkynyloxy, aryloxy, C₂-C₂₀ alkoxycarbonyl, C₁-C₂₀ alkylthio, C₁-C₂₀ alkylsulfonyl, C₁-C₂₀ alkylsulfinyl; each optionally substituted with C₁-C₅ alkyl, halogen, C₁-C₅ alkoxy or with a phenyl group optionally substituted with halogen, C₁-C₅ alkyl or C₁-C₅ alkoxy.
5. A compound of Claim 1 wherein:
- R and R¹ are independently selected from hydrogen; vinyl, C₁-C₁₀ alkyl, aryl, C₁-C₁₀ carboxylate, C₂-C₁₀ alkoxycarbonyl, C₁-C₁₀ alkoxy or aryloxy; each optionally substituted with C₁-C₅ alkyl, halogen, C₁-C₅ alkoxy or with a phenyl group optionally substituted with halogen, C₁-C₅ alkyl or C₁-C₅ alkoxy;
- X and X¹ are independently selected from halogen, hydrogen; C₁-C₂₀ alkyl, aryl, C₁-C₂₀ alkoxide, aryloxide, C₃-C₂₀ alkyldiketonate, aryldiketonate, C₁-C₂₀ carboxylate, aryl or C₁-C₂₀ alkyl sulfonate, C₁-C₂₀ alkylthio, C₁-C₂₀ alkylsulfonyl or C₁-C₂₀ alkylsulfinyl; each optionally substituted with C₁-C₅ alkyl, halogen, C₁-C₅ alkoxy or with a phenyl group optionally substituted with halogen, C₁-C₅ alkyl or C₁-C₅ alkoxy;
- L and L¹ are independently selected from phosphine, sulfonated phosphine,

phosphite, phosphinite, phosphonite,
arsine, stibine, ether, amine, amide,
sulfoxide, carbonyl, nitrosyl, pyridine or
thioether.

5

6. A compound of Claim 5 wherein:

R and R¹ are independently selected from
hydrogen; vinyl, C₁-C₅ alkyl, phenyl,
C₂-C₅ alkoxy carbonyl, C₁-C₅ carboxylate,
C₁-C₅ alkoxy, phenoxy; each optionally
substituted with C₁-C₅ alkyl, halogen,
C₁-C₅ alkoxy or a phenyl group optionally
substituted with halogen, C₁-C₅ alkyl or
C₁-C₅ alkoxy;

15

X and X¹ are independently selected from Cl,
Br, H; or benzoate, C₁-C₅ carboxylate,
C₁-C₅ alkyl, phenoxy, C₁-C₅ alkoxy, C₁-C₅
alkylthio, aryl, C₁-C₅ or alkyl sulfonate;
each optionally substituted with C₁-C₅
alkyl or a phenyl group optionally
substituted with halogen, C₁-C₅ alkyl or
C₁-C₅ alkoxy;

20

L and L¹ are independently selected from aryl-
or C₁-C₁₀ alkylphosphine, aryl- or C₁-C₁₀
alkylsulfonated-phosphine, aryl- or C₁-C₁₀
alkylphosphinite, aryl- or C₁-C₁₀
phosphonite, aryl- or C₁-C₁₀ alkyl
phosphite, aryl- or C₁-C₁₀ alkylarsine,
aryl- or C₁-C₁₀ alkylamine, pyridine,
aryl- or C₁-C₁₀ alkylsulfoxide, aryl- or
C₁-C₁₀ alkylether, aryl- or C₁-C₁₀
alkylamide; each optionally substituted
with C₁-C₅ alkyl, halogen, C₁-C₅ alkoxy,
or with a phenyl group optionally

25

30

substituted with halogen, C₁-C₅ alkyl or C₁-C₅ alkoxy; or

X, L and L¹ optionally can be taken together to be cyclopentadienyl optionally substituted with hydrogen; vinyl, C₁-C₅ alkyl, phenyl, C₂-C₅ alkoxy carbonyl, C₁-C₅ carboxylate, C₁-C₅ alkoxy, phenoxy; each optionally substituted with C₁-C₅ alkyl, halogen, C₁-C₅ alkoxy or a phenyl group optionally substituted with halogen, C₁-C₅ alkyl or C₁-C₅ alkoxy.

7. A compound of Claim 6 wherein:

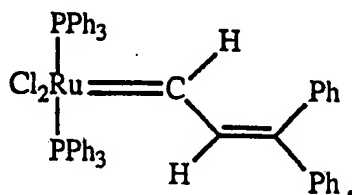
R and R¹ are independently, hydrogen, vinyl, Me, Ph, fluoroalkyl containing 1 to 5 carbon atoms;

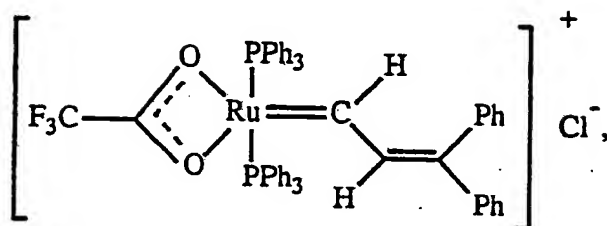
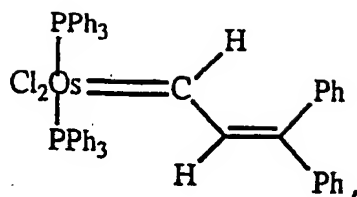
X and X¹ are independently Cl, CF₃CO₂, CH₃CO₂, C(F)H₂CO₂, (CH₃)₃CO, (CF₃)₂(CH₃)CO, (CF₃)(CH₃)₂CO, PhO, MeO, EtO, tosylate, mesylate or trifluoromethanesulfonate; and

L and L¹ are independently PMe₃, PPh₃, P(p-Tol)₃, P(o-Tol)₃, PMePh₂, PPhMe₂, P(CF₃)₃, P(p-FC₆H₄)₃, pyridine, P(p-CF₃C₆H₄)₃, (p-F)pyridine, (p-CF₃)pyridine, P(C₆H₄-SO₃Na)₃, or P(CH₂C₆H₄-SO₃Na)₃.

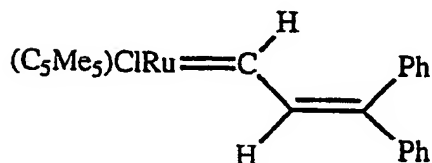
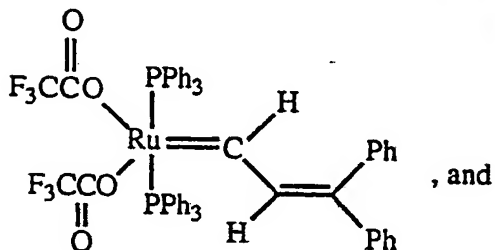
8. A compound selected from the group consisting

of:



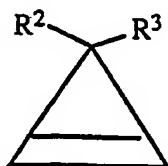


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9. A method of preparing compound of Claim 1 comprising reacting compound of the formula $(\text{XX}^1\text{ML}_n\text{L}^1_m)_p$, in the presence of solvent, with a cyclopropene of the formula



15

wherein:

M is Os or Ru;

X and X¹ are independently selected from any anionic ligand;

L and L¹ are independently selected from any neutral electron donor;

5 n and m are independently 0-4, provided n+m=2, 3 or 4;

p is an integer equal to or greater than 1; and

R² and R³ are independently selected from hydrogen;

10 C₁-C₁₈ alkyl, C₂-C₁₈ alkenyl, C₂-C₁₈ alkynyl, C₂-C₁₈ alkoxy carbonyl, aryl, C₁-C₁₈ carboxylate, C₁-C₁₈ alkenyloxy, C₂-C₁₈ alkynyloxy, C₁-C₁₈ alkoxy, aryloxy, C₁-C₁₈ alkylthio, C₁-C₁₈ alkylsulfonyl, C₁-C₁₈ alkylsulfinyl; each optionally substituted with C₁-C₅ alkyl,
15 halogen, C₁-C₅ alkoxy or with a phenyl group optionally substituted with halogen, C₁-C₅ alkyl or C₁-C₅ alkoxy.

20 10. A method according to Claim 9 wherein each said solvent is organic solvent.

11. A method according to Claim 9 conducted at a temperature of 25°C-125°C.

25 12. A method according to Claim 9 conducted under ultraviolet radiation.

30 13. A method according to Claim 9 conducted in the presence of a catalytic amount of HgCl₂ at a temperature between 15°C and 65°C.

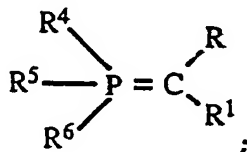
14. A method according to Claim 9 wherein:

X, L, and L¹ are taken together to be cyclopentadienyl, indenyl or fluorenyl
35 each optionally substituted with hydrogen;

5 C₂-C₂₀ alkenyl, C₂-C₂₀ alkynyl, C₁-C₂₀
alkyl, aryl, C₁-C₂₀ carboxylate, C₁-C₂₀
alkoxy, C₂-C₂₀ alkenyloxy, C₂-C₂₀
alkynyloxy, aryloxy, C₂-C₂₀
alkoxycarbonyl, C₁-C₂₀ alkylthio, C₁-C₂₀
alkylsulfonyl or C₁-C₂₀ alkylsulfinyl;
each optionally substituted with C₁-C₅
alkyl, halogen, C₁-C₅ alkoxy or with a
10 phenyl group optionally substituted with
halogen, C₁-C₅ alkyl or C₁-C₅ alkoxy.

15 15. A method according to Claim 14 wherein X, L,
and L¹ together are C₅Me₅; and M is Ru.

16. A method of preparing a compound of Claim 1
comprising reacting compound of the formula (XX¹ML_nL¹_m)_p
in the presence of solvent with phosphorane of the
formula



wherein:

- M is Ru or Os;
X and X¹ are independently selected from any anionic
25 ligand;
L and L¹ are independently selected from any neutral
electron donor;
n and m are independently 0-4, provided n+m is 2, 3
or 4;
30 p is an integer equal to or greater than 1;
R and R¹ are independently selected from hydrogen;
C₁-C₂₀ alkyl, C₂-C₂₀ alkenyl, C₂-C₂₀ alkynyl,

5 C₂-C₂₀ alkoxy carbonyl, aryl, C₁-C₂₀ carboxylate, C₂-C₂₀ alkenyloxy, C₂-C₂₀ alkynyloxy, C₁-C₂₀ alkoxy, aryloxy, C₁-C₂₀ alkylthio, C₁-C₂₀ alkylsulfonyl or C₁-C₂₀ alkylsulfinyl; each optionally substituted with C₁-C₅ alkyl, halogen, C₁-C₅ alkoxy or with a phenyl group optionally substituted with halogen, C₁-C₅ alkyl or C₁-C₅ alkoxy; and

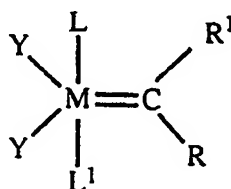
10 R⁴, R⁵, and R⁶ are independently selected from aryl, C₁-C₆ alkyl, C₁-C₆ alkoxy or phenoxy; each optionally substituted with halogen, C₁-C₃ alkyl, C₁-C₃ alkoxy, or with a phenyl group optionally substituted with halogen, C₁-C₅ alkyl or C₁-C₅ alkoxy.

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17. A method according to Claim 20 wherein any 2, 3 or 4 of X, X¹, L, L¹ optionally can be bonded together to create a chelating multidentate ligand.

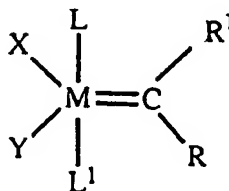
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18. A method of preparing compounds of Formulae II and III



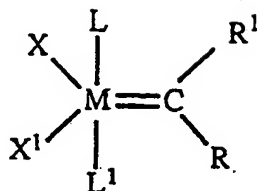
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II



III

from compound of Formula I



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I

comprising reacting said compound I, in the presence of solvent, with compound of the formula M^1Y wherein:

M is Os or Ru;

R and R^1 are independently selected from hydrogen;

10 C₂-C₂₀ alkenyl, C₂-C₂₀ alkynyl, C₁-C₂₀ alkyl, aryl, C₁-C₂₀ carboxylate, C₁-C₂₀ alkoxy, C₂-C₂₀ alkenyloxy, C₂-C₂₀ alkynyloxy, aryloxy, C₁-C₂₀ alkylthio, C₁-C₂₀ alkylsulfonyl or C₁-C₂₀ alkylsulfinyl; each optionally substituted with
 15 C₁-C₅ alkyl, halogen, C₁-C₅ alkoxy or with a phenyl group optionally substituted with halogen, C₁-C₅ alkyl or C₁-C₅ alkoxy;

X and X^1 are independently selected from any anionic ligand; and

20 L and L^1 are independently selected from any neutral electron donor;

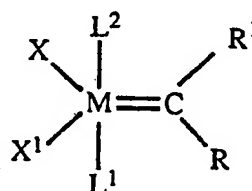
provided that

- (1) M^1 is Li, Na or K, and Y is C₁-C₁₀ alkoxide or arylalkoxide, each optionally substituted with
 25 C₁-C₁₀ alkyl or halogen, diaryloxy, or
 (2) M^1 is Na or Ag, and Y is ClO₄, PF₆, BF₄, SbF₆, halogen, B(aryl)₄, C₁-C₁₀ alkyl sulfonate or aryl sulfonate.

30 19. A method according to Claim 18 wherein any 2, 3 or 4 of X, X^1 , L, L^1 are bonded together to create a chelating multidentate ligand.

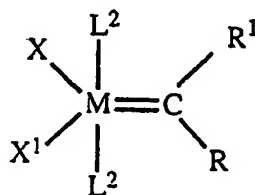
20. A method according to Claim 19 wherein X, L
and L¹ are taken together to be cyclopentadienyl,
indenyl or fluorenyl, each optionally substituted with
5 hydrogen; C₂-C₂₀ alkenyl, C₂-C₂₀ alkynyl, C₁-C₂₀ alkyl,
aryl, C₁-C₂₀ carboxylate, C₁-C₂₀ alkoxy, C₂-C₂₀
alkenyloxy, C₂-C₂₀ alkynyloxy, aryloxy, C₂-C₂₀
alkoxycarbonyl, C₁-C₂₀ alkylthio, C₁-C₂₀ alkylsulfonyl
or C₁-C₂₀ alkylsulfinyl; each optionally substituted
10 with C₁-C₅ alkyl, halogen, C₁-C₅ alkoxy or with a phenyl
group optionally substituted with halogen, C₁-C₅ alkyl
or C₁-C₅ alkoxy.

21. A method of preparing compounds of Formulae IV
15 and V



IV

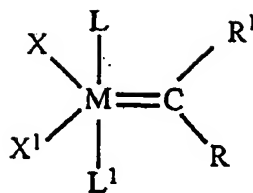
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V

25 from compound of Formula I

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I

5 comprising reacting said compound I, in the presence of solvent, with L² wherein:

M is Os or Ru;

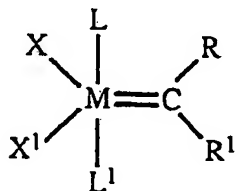
R and R¹ are independently selected from hydrogen;

10 C₂-C₂₀ alkenyl, C₂-C₂₀ alkynyl, C₁-C₂₀ alkyl, aryl, C₁-C₂₀ carboxylate, C₁-C₂₀ alkoxy, C₂-C₂₀ alkenyloxy, C₂-C₂₀ alkynyloxy, aryloxy, C₁-C₂₀ alkylthio, C₁-C₂₀ alkylsulfonyl or C₁-C₂₀ alkylsulfinyl; each optionally substituted with C₁-C₅ alkyl, halogen, C₁-C₅ alkoxy or with a
 15 phenyl group optionally substituted with halogen, C₁-C₅ alkyl or C₁-C₅ alkoxy;

X and X¹ are independently selected from any anionic ligand; and

20 L, L¹ and L² are independently selected from any neutral electron donor.

22. In the process of metathesis polymerization of cyclic olefin, the improvement comprising carrying out the polymerization in the presence of catalyst of the
 25 formula



in the presence of solvent, wherein:

M is Os or Ru;

R and R¹ are independently selected from hydrogen;
C₁-C₂₀ alkyl, C₂-C₂₀ alkenyl, C₂-C₂₀ alkynyl,
C₂-C₂₀ alkoxycarbonyl, aryl, C₁-C₂₀ carboxylate,
5 C₁-C₂₀ alkoxy, C₂-C₂₀ alkenyloxy, C₂-C₂₀
alkynyloxy or aryloxy; each optionally
substituted with C₁-C₅ alkyl, halogen, C₁-C₆
alkoxy or with a phenyl group substituted with
halogen, C₁-C₅ alkyl or C₁-C₅ alkoxy;

10 X and X¹ are independently selected from any anionic
ligand; and

L and L¹ are independently selected from any neutral
electron donor.

15 23. In the process of Claim 22, wherein the cyclic
olefin is norbornene, norbornadiene, cyclopentene,
dicyclopentadiene, cycloheptene, cyclo-octene,
7-oxanorbornene, 7-oxanorbornadiene, and cyclododecene.

20 24. A process according to Claim 22 wherein the
catalyst is dissolved in protic, or aqueous solvent or
mixture of aqueous, protic and/or organic solvents.

25 25. A process according to Claim 22 wherein any
2,3 or 4 of X, X¹, L, L¹ are bonded together to create a
chelating multidentate ligand.

30 26. A process according to Claim 25 wherein X, L
and L¹ are taken together to be cyclopentadienyl,
indenyl or fluorenyl, each optionally substituted with
hydrogen; C₂-C₂₀ alkenyl, C₂-C₂₀ alkynyl, C₁-C₂₀ alkyl,
aryl, C₁-C₂₀ carboxylate, C₁-C₂₀ alkoxy, C₂-C₂₀
alkenyloxy, C₂-C₂₀ alkynyloxy, aryloxy, C₂-C₂₀
alkoxycarbonyl, C₁-C₂₀ alkylthio, C₁-C₂₀ alkylsulfonyl
35 or C₁-C₂₀ alkylsulfinyl; each optionally substituted

with C₁-C₅ alkyl, halogen, C₁-C₅ alkoxy or with a phenyl group optionally substituted with halogen, C₁-C₅ alkyl or C₁-C₅ alkoxy.

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